acid reductase. However, adaptive increased specific activities of two hepatic tetrahydrofolate-dependent enzymes, formyltetrahydrofolic acid synthetase and formiminotetrahydrofolic acid transferase, could not be demonstrated.

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REFERENCES

- 1. J. R. BERTINO, Cancer Res. 23, 1286 (1963).
- 2. R. C. Wood, R. Ferone and G. H. HITCHINGS, Biochem. Pharmac. 6, 113 (1961).
- 3. G. A. FISCHER, Biochem. Pharmac. 11, 1233 (1962).
- 4. G. A. FISCHER, Biochem. Pharmac. 7, 75 (1961).
- 5. D. K. Misra, S. R. Humphreys, M. Friedkin, A. Goldin and E. J. Crawford, *Nature, Lond.* 189, 39 (1961).
- 6. A. M. Albrecht, F. K. Pearce and D. J. Hutchinson, J. biol. Chem. 241, 1036 (1966).
- 7. O. H. LOWRY, N. J. ROSENBROUGH, A. L. FARR and R. J. RANDALL, J. biol. Chem. 193, 265 (1951).
- 8. R. J. HENRY, in Clinical Chemistry Principles and Technics, p. 182, Harper, New York (1964).
- 9. R. L. Blakley, The Biochemistry of Folic Acid and Related Pteridines, p. 94, North-Holland Amsterdam (1969).
- 10. H. TABOR and L. WYNGARDEN, J. biol. Chem. 234, 1830 (1959).
- 11. S. FUTTERMAN, J. biol. Chem. 228, 1031 (1957).
- 12. B. L. HILLCOAT, P. F. NIXON and R. L. BLAKLEY, Analyt. Biochem. 21, 178 (1967).
- 13. J. R. Bertino, B. W. Gabrio and F. M. Huennekens, Biochem. biophys. Res. Commun. 3, 461 (1960).
- 14. M. F. MARTELLI, M. TONATO and F. GRINGNANI, Enzym. biol. clin. 8, 353 (1967).
- 15. C. K. Mathews and F. M. Huennekens, J. biol. Chem. 238, 3436 (1963).
- J. R. Bertino, D. M. Donohue, B. Simmons, B. W. Gabrio, R. Silber and F. M. Huennekens, J. clin. Invest. 42, 466 (1963).
- 17. J. R. BERTINO, B. SIMMONS and D. M. DONOHUE, Biochem. Pharmac. 13, 225 (1964).
- 18. A. C. SARTORELLI, B. A. BOOTH and J. R. BERTINO, Archs Biochem. Biophys. 108, 53 (1964).
- 19. W. C. WERKHEISER, J. biol. Chem. 236, 888 (1961).

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In vivo effects of carbon tetrachloride and chloroform on liver and kidney glucose-6-phosphatase in mice

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ALTHOUGH biochemical changes in the liver following the administration of carbon tetrachloride and chloroform are wellknown, ¹⁻⁸ little is known about biochemical changes occurring in the kidney. We have compared the effects of CHCl₃ and CCl₄ on glucose-6-phosphatase activity in mouse liver and kidney.

Experimental

Animals. CFLP mice were obtained from Carworth Europe, Alconbury, Hunts; TF1 mice from A. Tuck, Rayleigh, Essex and CBA mice from the Laboratory Animal Centre, Carshalton, Surrey. They were used when they were between 20 and 30 g in weight.

Treatment. The pelleted diet FFG (Dixon & Son, Ware, Herts) and water were available ad lib. CCl₄ and CHCl₃ were given orally as 50% (v/v) solutions in liquid paraffin. Animals were killed by cervical dislocation 21 hr after dosing.

Enzyme preparation. Livers or kidneys were homogenized in 9 vol. of ice-cold 0.25 M sucrose-0.1 M cacodylate-0.001 M EDTA buffer (pH 7.0) in a Teflon-glass homogeniser. The homogenate was filtered through surgical gauze and the filtrate was used directly for enzyme assay.

Enzyme assay. The enzyme incubation mixture contained 0.4 ml of 0.1 M sodium cacodylate buffer (pH 6.1), 0.5 ml of 0.05 M disodium glucose-6-phosphate (Wessex Biochemicals, Bournemouth) and 0.1 ml homogenate. After 15 min incubation at 30°, the reaction was terminated by the addition of 2 ml trichloroacetic acid. Inorganic phosphate was determined in a 1-ml aliquot of protein-free supernatant by the method of Fiske and Subbarow.⁹

Protein concentration. This was measured by the method of Lowry et al. 10

Liver lipid. This was determined by application of the methods of Folch et al.¹¹ and Bligh and Dyer¹² as described by Diplock et al.¹³

Results

Comparative effects of CCl₄ and CHCl₃ on liver glucose-6-phosphatase in male and female mice. The results are shown in Table 1. CCl₄ produced a significant fall in liver glucose-6-phosphatase in both sexes but CHCl₃ had no effect. These results agree with previous findings.^{1,2,8}

Comparative effects of CCl₄ and CHCl₃ on kidney glucose-6-phosphatase in male and female mice. The results are shown in Table 2. Both CCl₄ (2 ml/kg) and CHCl₃ (1 ml/kg) decreased kidney glucose-6-phosphatase activity in male mice, but had little effect in female mice.

Sex	Halohydrocarbon (2 ml/kg)	Glucose-6-phosphatase (μ mole PO ₄ /g liver/hr)		Chausa
		Control	Treated	Change (%)
M	CCl₄	981 ± 28	711 ± 72†	-28
F	CCl ₄	987 ± 55	$651 \pm 69^{+}$	-35
M	CHCl ₃	1205 ± 56	1165 ± 89	-3
F	CHCl ₃	1028 ± 22	1227 ± 92	+19

TABLE 1. EFFECTS OF CCl4 AND CHCl3 ON LIVER GLUCOSE-6-PHOSPHATASE*

Effect of CHCl₃ on liver lipid accumulation. Although CHCl₃ did not affect liver glucose-6-phosphatase it did, like CCl₄, increase liver lipids. We found an increase in the neutral lipid content from (mean \pm S.D.) $28\cdot1$ \pm 8·7 mg/g to $48\cdot5$ \pm 10·6 in male CFLP mice, from $31\cdot8$ \pm 1·6 to $53\cdot7$ \pm 7·0 in female CFLP mice, from $14\cdot2$ \pm 1·6 to $44\cdot0$ \pm 16·5 in male TF1 mice and from $22\cdot2$ \pm 1·7 to $50\cdot4$ \pm 8·3 in female TF1 mice.

Discussion

The administration of CCl₄ to animals causes, within a few hours, a loss of glucose-6-phosphatase^{1,2} and, within 24 hr fatty liver and hepatic necrosis. The toxicity of CCl₄ has been shown to be dependent on its metabolism via a toxic intermediate to CO₂.¹⁴ After CCl₄ is given to rats, the presence in microsomal lipids of a material having the optical characteristics of a conjugated diene has been demonstrated.¹⁵ Pretreatment of rats with phenobarbitone, although increasing the toxicity of CCl₄,¹⁴ prevents the CCl₄-induced loss of glucose-6-phosphatase.¹⁶

CHCl₃ also produces fatty liver and necrosis but, unlike CCl₄, does not affect either hepatic glucose-6-phosphatase or produce detectable amounts of diene in liver. In our experiments the *in vivo* effects of CCl₄ and CHCl₃ on hepatic glucose-6-phosphatase have been confirmed. In addition, however,

^{*} Results are the mean \pm S.E. of three individual liver analyses.

[†] Significantly lower than controls; P < 0.05.

Table 2. The effects of oral CCl₄ (2.0 ml/kg) and CHCl₃ (1.0 ml/kg) on kidney glucose-6-phosphatase*

Strain	Sex	Halohydrocarbon	Kidney glucose-6-phosphatase (μmole PO ₄ /100 mg protein/hr)	
			Control	Treated
CFLP	М	CCl ₄	613 ± 59 (3)	483 ± 25 (3)
CFLP	F	CCl ₄	$768 \pm 48 (3)$	$757 \pm 81 (3)$
CFLP	M	CHCl ₃	$621 \pm 22 (9)$	$392 \pm 18 (15) \dagger$
CFLP	F	CHCl ₃	$699 \pm 27 (4)$	$588 \pm 43 (7)$
TF1	M	CHCl ₃	$589 \pm 37 (4)$	$339 \pm 15 (8) \dagger$
TF1	F	CHCl ₃	$482 \pm 40 \ (4)$	$494 \pm 21 \ (10)$
CBA	M	CHCl ₃	$501 \pm 17 (2)$	$228 \pm 7(5)^{\dagger}$
CBA	F	CHCl ₃	463 (1)	$424 \pm 20 (2)$

^{*} Mice were killed 21 hr after dosage and kidneys excised. Results are given as means \pm S.E. with the no. of mice in parentheses.

it has been shown that both halohydrocarbons produce loss of kidney glucose-6-phosphatase in mice and this effect is sex-dependent, being greatest in males. These results demonstrate that hepatic fat accumulation and loss of glucose-6-phosphatase cannot be directly due to a common primary lesion and, when coupled with previous studies¹⁷ on the protective action of antioxidants against CCl₄ toxicity, reinforce the view⁸ that "changes seen with glucose-6-phosphatase and lipoperoxidation after CCl₄ are only secondary effects and have little to do with the initiating event in the development of the lesion".

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REFERENCES

- 1. R. O. RECKNAGEL and B. LOMBARDI, J. biol. Chem. 236, 564 (1961).
- 2. E. S. REYNOLDS and A. G. YEE, Lab. Invest. 19, 273 (1968).
- 3. J. V. DINGELL and L. E. FRANKLIN, Fedn Proc. Fedn Am. Socs. exp. Biol. 26, 354 (1967).
- 4. D. NEUBERT and D. MAIBAUER, Arch. exp. Path. Pharmak. 235, 291 (1959).
- 5. R. KATO, E. CHIESARA and P. VASSANELLI, Biochem. Pharmac. 11, 211 (1962).
- 6. C. D. KLAASSEN and G. L. PLAA, Toxic. appl. Pharmac. 9, 139 (1966).
- 7. C. D. KLAASSEN and G. L. PLAA, Toxic. appl. Pharmac. 10, 119 (1967).
- 8. C. D. KLAASSEN and G. L. PLAA, Biochem. Pharmac. 18, 2019 (1969).
- 9. C. H. FISKE and Y. SUBBAROW, J. biol. Chem. 66, 375 (1925).
- 10. O. H. LOWRY, N. J. ROSEBROUGH, A. L. FARR and R. J. RANDALL, J. biol. Chem. 193, 265 (1951).
- 11. J. Folch, M. Lees and G. H. S. Stanley, J. biol. Chem. 226, 497 (1957).
- 12. E. G. BLIGH and W. J. DYER, Can. J. Biochem. Physiol. 37, 911 (1959).
- A. T. DIPLOCK, M. A. CAWTHORNE, E. A. MURRELL, J. GREEN and J. BUNYAN, Br. J. Nutr. 22, 465 (1968).
- 14. R. C. GARNER and A. E. M. McLean, Biochem. Pharmac. 18, 645 (1969).
- 15. R. O. RECKNAGEL and A. K. GHOSHAL, Expl. Mol. Path. 5, 413 (1966).
- 16. G. FEUER, L. GOLBERG and A. HUNT, Biochem. J. 102, 7P (1967).
- M. A. CAWTHORNE, J. BUNYAN, M. V. SENNITT, J. GREEN and P. GRASSO, Br. J. Nutr. 24, 357 (1970).

[†] Significantly lower than control value: P < 0.001.